Atty. Docket No.: AFIP03-58 Amdt. Dated January 31, 2008

Reply to Office action of October 19, 2007

Appl. No.: 10/827,282

## Amendments to the Claims:

This listing of claims will replace all prior versions, and listings, of claims in the application:

## Listing of Claims:

1. (previously presented) A method of detecting oxidants in a biological sample comprising:

adding a source of ferrous ions to said sample, whereby the presence of oxidants in said sample oxidize at least a portion of said ferrous ions to ferric ions;

adding a chromogenic compound to said sample, whereby said chromogenic compound reacts with at least a portion of any ferric ions present in said sample; and

detecting for the product of said chromogenic compound-ferric ion reaction; whereby the detection of said chromogenic compound-ferric ion reaction product indicates the presence of oxidants in said sample.

- (previously presented) The method of claim 1 wherein said sample is a biological sample.
- 3. (previously presented) The method of claim 2 wherein said sample is urine.
- 4. (previously presented) The method of claim 1 wherein said source of ferrous ions is ferrous ammonium sulfate.
- 5. (previously presented) The method of claim 1 wherein said oxidants are selected from the group consisting of ferric, chromates, permanganates, iodates, periodates, oxychlorides, hydroperoxides, hydrogen peroxides, persulfates, oxone, tert-butyl hydrogen peroxide, cumene hydrogen peroxide, and nitrites.
- (previously presented) The method of claim 1 wherein said chromogenic compound-ferric ion reaction product is a chromogen from chromogenic oxidation of said chromogenic compound.

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7. (previously presented) The method of claim 6, wherein said chromogenic compound is selected from the group consisting of 2,2'-Azino-bis(3-ethylbenzthiazoline-6-sulfonic acid), N,N-Dimethylphenylenediamine, and 2-Amino-p-cresol.

- 8. (previously presented) The method of claim 7 wherein said chromogen is detected visually.
- (previously presented) The method of claim 7 wherein said chromogen is detected via spectrophotometric analysis.
- 10. (previously presented) The method of claim 4 wherein said ferrous ions are the product of dissolving ferrous ammonium sulfate in hydrochloric acid in the presence of vanadium.
- 11. (previously presented) The method of claim 1 wherein said chromogenic compound-ferric ion reaction product is a chromogenic complex.
- 12. (previously presented) The method of claim 11 wherein said chromogenic compound is selected from the group consisting of Xylenol orange, 8-Hydroxy-7-iodo-5-quinolinesulfonic acid, and 4,5-Dihydroxy-1,3-benzene-di-sulfonic acid.
- 13. (previously presented) The method of claim 12 wherein said chromogenic complex is detected visually.
- 14. (previously presented) The method of claim 12 wherein said chromogenic complex is detected via spectrophotometric means.
- 15. (previously presented) The method of claim 12 wherein said ferrous ion is present in excess.
- 16. (previously presented) The method of claim 1 wherein said detecting step comprises detecting a concentration of said reaction product.
- 17. (previously presented) The method of claim 16 wherein said concentration is determined by comparing an intensity of said sample to intensities of standards having

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known concentrations of oxidants.

18. (previously presented) The method of claim 17 wherein said comparison is performed using spectrophotometric means.

19. (previously presented) A method of detecting adulteration of a urine sample comprising:

adding a source of ferrous ions to a urine sample;

adding a chromogenic compound to said urine sample;

detecting the presence or absence of a chromogenic reaction product;

determining a concentration of said chromogenic reaction product; and

determining if said concentration signifies adulteration of said urine sample.

 (previously presented) The method of claim 19 wherein said source of ferrous ions is ferrous ammonium sulfate.

- 21. (previously presented) The method of claim 19 wherein said chromogenic compound is selected from the group consisting of 2,2'-Azino-bis(3-ethylbenzthiazoline-6-sulfonic acid), N,N-Dimethylphenylenediamine, and 2-Amino-p-cresol, and wherein said chromogenic reaction product is a chromogenic oxidation product.
- 22. (previously presented) The method of claim 19 wherein said chromogenic compound is selected from the group consisting of Xylenol orange, 8-Hydroxy-7-iodo-5-quinolinesulfonic acid, and 4,5-Dihydroxy-1,3-benzene-di-sulfonic acid, and wherein said chromogenic reaction product is a chromogenic complex.
- 23. (previously presented) The method of claim 19 wherein said chromogenic reaction product is detected visually.
- 24. (previously presented) The method of claim 19 wherein said concentration of chromogenic reaction product is determined spectrophotometrically.

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25. (previously presented) The method of claim 24 wherein said concentration is determined in milliequivalents per liter of oxidants.

- 26. (previously presented) The method of claim 25 wherein said milliequivalents per liter is based on comparison of said spectrophotometric results with standards having known milliequivalents per liter.
- 27. (previously presented) The method of claim 26 wherein said adulteration determination is based on comparison to an oxidizing property of unadulterated urine.
- 28. (previously presented) The method of claim 27 wherein said sample is determined to be adulterated if the oxidant concentration is determined to exceed 29 milliequivalents per liter.